Structural distortions in Pd–MeO-Biphep and Pd–Binap aryl complexes. Anomalies induced *via* electronic effects

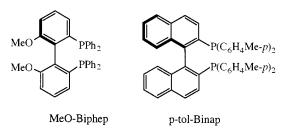
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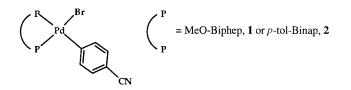
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The four-coordinate complexes $[PdBr(p-NCC_6H_4)-(MeO-Biphep)]$, 1 [MeO-Biphep = 6,6'-dimethoxy-2,2'-bis-(diphenylphosphino)-1,1'-biphenyl], and $[PdBr(p-NC-C_6H_4)(p-tol-Binap)]$, 2 [p-tol-Binap = 2,2'-bis(di-p-tolyl-phosphino)-1,1'-binaphthyl], both distort markedly, with the former no longer square-planar, and the latter, possessing the longest recorded Pd–P bond length.

The atropisomeric chiral bidentates MeO-Biphep and Binap are



recognised to be excellent auxiliaries in a number of enantioselective transformations,¹⁻³ including C–C coupling.⁴ Pd–aryl complexes, which are intermediates in both cross-coupling and Heck chemistry,⁴ arise from the oxidative addition of aryl–X compounds, X = I, Br, OTs, to Pd(0) complexes.⁵⁻¹⁰ In connection with kinetic studies related to the enantioselective Heck reaction we recently noted¹¹ that aryl intermediates of the type [PdX(aryl)(MeO-Biphep)], with electron-donor groups on the aryl, are not very stable. We report here that the structures of the compounds [PdBr(*p*-NCC₆H₄)(MeO-Biphep)], **1**, and [PdBr(*p*-NCC₆H₄)(*p*-tol-Binap)], **2**, which are more stable, both distort markedly (but very differently) in order to accommodate the electronic pressure exerted by the aryl groups,†



The molecular structures for 1 and 2 were determined *via* X-ray diffraction methods \ddagger and ORTEP views of these molecules are given in Figs. 1 and 2. Selected bond distances and bond angles are given in the captions. In compound 1 there is sufficient space for the aryl and bromide ligands; nevertheless both ligands deviate so strongly from the coordination plane defined by the two P-donors and the metal, +0.88 Å and -0.57 Å, respectively, that one can no longer speak of a square planar geometry. We envision the structure as arising *via* rotation of the P–Pd–P and Cl–Pd–Br, planes, relative to one another, and not due to a tetrahedral distortion (note that the P–Pd–P and C1L–Pd–Br angles are *ca*. 94° and 90°, respectively). The Pd–C1L separation of 2.104(3) Å is relatively long [2.04–2.06 Å is normal for MeO-Biphep¹¹ although, in general much shorter Pd–C(aryl) distances have been reported ^{12,13}]. The ring of the



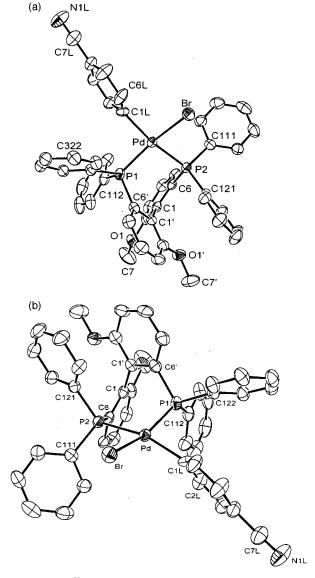


Fig. 1 ORTEP²⁰ views of complex **1**. (a) From above and (b) from behind (and slightly above) the *p*-cyanoaryl ligand. Selected bond lengths (Å) and bond angles (°) for **1**: Pd–P1 2.2700(9), Pd–P2 2.3501(9), Pd–C1L 2.104(3), Pd–Br 2.4920(4), C7L–N1 1.126(5); P1–Pd–P2 94.06(3), P2–Pd–C1L 163.02(10), P1–Pd–C1L 92.22(9), P1–Pd–Br 158.96(3), P2–Pd–Br 89.69(2), C1L–Pd–Br 90.09(9).

p-NCC₆H₄ ligand makes an angle of *ca*. 73° with the P–Pd–P plane, but seems to be somewhat bent, P2–Pd–C1L = 163°, see Fig. 1a. We find no evidence for strain due to packing effects.

For the Binap complex 2 the observed coordination geometry is slightly distorted square planar, see Fig. 2. Relative to 1 the Pd–C1L bond separation is now 0.078 Å shorter, 2.026(6) Å;¹⁴ however, a search of the Cambridge database reveals that the

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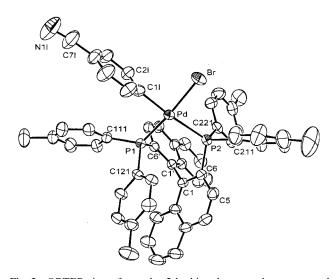
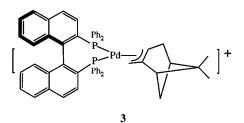


Fig. 2 ORTEP view of complex 2 looking down on the *p*-cyanoaryl ligand and the coordination plane. Selected bond lengths (Å) and bond angles (°) for 2: Pd-P1 2.254(1), Pd-P2 2.437(1), Pd-C1L, 2.026(6), Pd-Br 2.4484(7), C7-N1 1.12(1); P1-Pd-P2 92.38(5), P2-Pd-C1L 175.6(2), P1-Pd-C1L 91.0(2), P1-Pd-Br 173.51(4), P2-Pd-Br 90.90(4), C1L-Pd-Br 85.4(2).

Pd-P2 bond, trans to the aryl, at 2.437(1) Å represents the longest Pd-P bond ever reported.§ For comparison, in the Binap β -pinene allyl complex $[Pd(\eta^3 - C_{10}H_{15})\{(R) - (+) - Binap\}]$ -[CF₃SO₃], 3, the Pd–P distances are 2.312(3) Å and 2.347(5)



Å.15 The observed Pd-P2 bond in 1, 2.3501(9) Å, is much shorter, but lies towards the upper end of the literature range (Pd–P separations of the order of *ca.* 2.20–2.36 Å are common^{16–19}). In both 1 and 2 the *p*-CN group appears to be a normal triple bond (see captions).

Despite the superficial similarity of the ligands, i.e., biarylbased triaryl phosphine types, the structures for 1 and 2 are very different. Whereas the extreme lengthening of the Pd-P2 bond in the Binap complex 2 arises due to good donor properties of the aryl group, the MeO-Biphep analog 1 avoids this electronic strain by strongly deviating from square planar geometry and thus weakening the Pd-C1L overlap. Given the rather novel structural results for these p-NCC₆H₄ aryl compounds, it is not surprising that we cannot readily isolate analogous complexes with aryl ligands which are even more electron donating, e.g. p-MeC₆H₄ and p-MeOC₆H₄.

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Notes and references

[†] The complexes were prepared as described in reference 11. ¹ Crystal data for compound 1: C₄₅H₃₆BrNO₂P₂Pd, M = 871.00, orthorhombic, space group $P2_{12}_{12}_{12}$ (no. 18), a = 11.4485(2), b = 17.3604(1), c = 19.0940(3) Å, U = 3794.94(9) Å³, Z = 4, $\mu = 16.66$ cm⁻¹, T = 200 K, R1 = 0.0267 (for 5794 unique reflections having $I > 2\sigma(I)$), 0.0354 (for all 6492 independent reflections). Crystal data for compound 2·CH₂Cl₂: C₅₆H₄₆BrCl₂NP₂Pd, M = 1052.16, monoclinic, space group C2 (no. 5), a = 29.0943(4), b = 11.8671(2), c = 17.2953(2) Å, $\beta = 116.790(3)^\circ$, U = 5330.51(13) Å³, Z = 4, $\mu = 12.94$ cm⁻¹, T = 293 K, R1 = 0.0419 (for 5550 unique reflections with $I > 2\sigma(I)$), 0.0494 (for all 7049 independent reflections). CCDC reference number 186/1478. § Note added at proof: a Pd-P bond length of ca. 2.5 Å (trans to SiCl₃) has been observed.21

- 1 R. Noyori, Chimia, 1988, 42, 215; R. Noyori, Asymmetric Catalysis in Organic Synthesis, John Wiley and Sons, New York, 1994
- 2 Y. Crameri, J. Foricher, U. Hengartner, C. Jenny, F. Kienzle, H. Ramuz, M. Scalone, M. Schlageter, R. Schmid and S. Wang, *Chimia*, 1997, **51**, 303; R. Schmid, M. Cereghetti, B. Heiser, P. Schönholzer and H. J. Hansen, *Helv. Chim. Acta*, 1988, **71**, 897; R. Schmid, E. A. Broger, M. Cereghetti, Y. Crameri, J. Foricher, M. Lalonde, R. K. Mueller, M. Scalone, G. Schoettel and U. Zutter, Pure Appl. Chem., 1996, 68, 131; C. Bolm, D. Kaufmann, S. Gessler and K. Harms, J. Organomet. Chem., 1995, 502, 47.
- 3 G. Trabesinger, A. Albinati, N. Feiken, R. W. Kunz, P. S. Pregosin and M. Tschoerner, J. Am. Chem. Soc., 1997, 119, 6315. 4 L. E. Overman and J. T. Link, in Metal Catalyzed Cross-Coupling
- Reactions, ed. F. A. S. Diederich, P. J., Weinheim, 1998.
- 5 A. de Meijere and F. E. Meyer, Angew. Chem., 1994, 106, 2473; C. Amatore, E. Carré, A Jutand, M. A. M'Barki and G. Meyer, Organometallics, 1995, 14, 5605.
- 6 P. Garrou and R. F. Heck, J. Am. Chem. Soc., 1976, 98, 4115; R. F. Heck, Acc. Chem. Res., 1979, **12**, 146; R. F. Heck, Comprehen-sive Organic Synthesis, ed. B. M. Trost and I. Flemming, Pergamon, Oxford, 1991
- 7 O. Loiseleur, P. Meier and A. Pfaltz, Angew. Chem., 1996, 108, 218.
- 8 F. Ozawa, A. Kubo and T. Hayashi, J. Am. Chem. Soc., 1991, 113, 1417
- 9 M. Tschoerner, G. Trabesinger, A. Albinati and P. S. Pregosin, Organometallics, 1997, 16, 3447.
- 10 J. F. Hartwig, Angew. Chem., Int. Ed., 1998, 37, 2090.
- 11 M. Tschoerner, P. S. Pregosin and A. Albinati, Organometallics, 1999, 18, 670.
- 12 A. G. Orpen, L. Brammer, F. H. Allen, O. Kennard, D. G. Watson and R. Taylor, J. Chem. Soc., Dalton Trans., 1989, S1; W. J. Marshall, D. L. Thorn and V. V. Grushin, Organometallics, 1998, 17, 5427
- 13 J. M. Brown, J. Perez-Torrente, N. Alcock and H. J. Clase, Organometallics, 1995, 14, 207. The Pd-C distance in the aryl complex PdI-{[2-(CH₂OCH=CH₂)C₆H₄](dppf)} is 2.055(7) Å.
- 14 J. M. Brown and J. J. P. Torrente, Organometallics, 1995, 14, 1195. Pt-C separations of 2.090(31) Å and 2.084(40) Å are reported for a biaryl-Binap complex. The Pt-P bond distances are 2.303(8) Å and 2.301(8) Å.
- 15 P. S. Pregosin, H. Rüegger, R. Salzmann, A. Albinati, F. Lianza and R. W. Kunz, Organometallics, 1994, 13, 83.
- 16 J. M. Wisner, T. J. Bartczak and J. A. Ibers, Organometallics, 1986, 5, 2044.
- 17 W. A. Herrmann, C. Brossmer, T. Priermeier and K. Oefele, J. Organomet. Chem., 1994, 481, 97
- 18 G. Mann, D. Baranano, J. F. Hartwig, A. L. Rheingold and I. A. Guzei, J. Am. Chem. Soc., 1998, 120, 9205.
- 19 D. K. Wicht, M. A. Zhuravel, R. V. Gregush, D. S. Glueck, I. A. Guzei, L. M. Liable-Sands and A. L. Rheingold, Organometallics, 1998, 17, 1412.
- 20 C. K. Johnson, ORTEP, Report ORNL-5138, Oak Ridge National Laboratory, Oak Ridge, TN, 1976.
- 21 A. Togni, personal communication.

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