# Structural distortions in Pd-MeO-Biphep and Pd-Binap aryl complexes. Anomalies induced via electronic effects 

Daniela Drago, ${ }^{a}$ Paul S. Pregosin, ${ }^{* a}$ Matthias Tschoerner ${ }^{a}$ and Alberto Albinati ${ }^{* b}$<br>${ }^{a}$ Laboratory of Inorganic Chemistry, ETH Zentrum, 8092 Zürich, Switzerland<br>${ }^{b}$ Chemical Pharmacy, University of Milan, I-20131 Milan, Italy

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The four-coordinate complexes $\left[\operatorname{PdBr}\left(p-\mathrm{NCC}_{6} \mathrm{H}_{4}\right)\right.$ -(MeO-Biphep)], 1 [MeO-Biphep = 6,6'-dimethoxy-2,2'-bis-(diphenylphosphino)-1,1'-biphenyl], and $[\operatorname{PdBr}(p-N C-$ $\left.\mathrm{C}_{6} \mathrm{H}_{4}\right)\left(p\right.$-tol-Binap)], 2 [p-tol-Binap $=2,2^{\prime}$-bis $($ di- $p$-tolyl-phosphino)-1,1'-binaphthyl], both distort markedly, with the former no longer square-planar, and the latter, possessing the longest recorded $\mathrm{Pd}-\mathrm{P}$ bond length.

The atropisomeric chiral bidentates MeO -Biphep and Binap are


MeO-Biphep

p-tol-Binap
recognised to be excellent auxiliaries in a number of enantioselective transformations, ${ }^{1-3}$ including C-C coupling. ${ }^{4}$ Pd-aryl complexes, which are intermediates in both cross-coupling and Heck chemistry, ${ }^{4}$ arise from the oxidative addition of aryl-X compounds, $\mathrm{X}=\mathrm{I}, \mathrm{Br}, \mathrm{OTs}$, to $\mathrm{Pd}(0)$ complexes. ${ }^{5-10}$ In connection with kinetic studies related to the enantioselective Heck reaction we recently noted ${ }^{11}$ that aryl intermediates of the type [PdX(ary))(MeO-Biphep)], with electron-donor groups on the aryl, are not very stable. We report here that the structures of the compounds $\left[\operatorname{PdBr}\left(p-\mathrm{NCC}_{6} \mathrm{H}_{4}\right)(\mathrm{MeO}-\mathrm{Biphep})\right]$, 1, and $\left[\operatorname{PdBr}\left(p-\mathrm{NCC}_{6} \mathrm{H}_{4}\right)(p\right.$-tol-Binap)], 2, which are more stable, both distort markedly (but very differently) in order to accommodate the electronic pressure exerted by the aryl groups. $\dagger$


The molecular structures for $\mathbf{1}$ and $\mathbf{2}$ were determined via X-ray diffraction methods $\ddagger$ and ORTEP views of these molecules are given in Figs. 1 and 2. Selected bond distances and bond angles are given in the captions. In compound $\mathbf{1}$ there is sufficient space for the aryl and bromide ligands; nevertheless both ligands deviate so strongly from the coordination plane defined by the two P-donors and the metal, $+0.88 \AA$ and -0.57 $\AA$, respectively, that one can no longer speak of a square planar geometry. We envision the structure as arising via rotation of the $\mathrm{P}-\mathrm{Pd}-\mathrm{P}$ and $\mathrm{Cl}-\mathrm{Pd}-\mathrm{Br}$, planes, relative to one another, and not due to a tetrahedral distortion (note that the $\mathrm{P}-\mathrm{Pd}-\mathrm{P}$ and C1L- $\mathrm{Pd}-\mathrm{Br}$ angles are $c a .94^{\circ}$ and $90^{\circ}$, respectively). The $\mathrm{Pd}-$ C1L separation of $2.104(3) \AA$ is relatively long [2.04-2.06 $\AA$ is normal for MeO-Biphep ${ }^{11}$ although, in general much shorter $\mathrm{Pd}-\mathrm{C}$ (aryl) distances have been reported ${ }^{12,13}$ ]. The ring of the


Fig. 1 ORTEP ${ }^{20}$ views of complex 1. (a) From above and (b) from behind (and slightly above) the $p$-cyanoaryl ligand. Selected bond lengths ( $\AA$ ) and bond angles $\left({ }^{\circ}\right)$ for 1: Pd-P1 2.2700(9), Pd-P2 2.3501(9), Pd-C1L 2.104(3), Pd-Br 2.4920(4), C7L-N1 1.126(5); P1-Pd-P2 94.06(3), P2-Pd-C1L 163.02(10), P1-Pd-C1L 92.22(9), P1-Pd- $\mathrm{Br} 158.96(3), \mathrm{P} 2-\mathrm{Pd}-\mathrm{Br} 89.69(2), \mathrm{C} 1 \mathrm{~L}-\mathrm{Pd}-\mathrm{Br} 90.09(9)$.
$p-\mathrm{NCC}_{6} \mathrm{H}_{4}$ ligand makes an angle of $c a .73^{\circ}$ with the $\mathrm{P}-\mathrm{Pd}-\mathrm{P}$ plane, but seems to be somewhat bent, $\mathrm{P} 2-\mathrm{Pd}-\mathrm{C} 1 \mathrm{~L}=163^{\circ}$, see Fig. 1a. We find no evidence for strain due to packing effects.
For the Binap complex 2 the observed coordination geometry is slightly distorted square planar, see Fig. 2. Relative to $\mathbf{1}$ the $\mathrm{Pd}-\mathrm{C} 1 \mathrm{~L}$ bond separation is now $0.078 \AA$ shorter, 2.026(6) $\AA \mathrm{F}^{14}$ however, a search of the Cambridge database reveals that the


Fig. 2 ORTEP view of complex 2 looking down on the $p$-cyanoaryl ligand and the coordination plane. Selected bond lengths ( $\AA$ ) and bond angles ( ${ }^{\circ}$ ) for 2: Pd-P1 2.254(1), Pd-P2 2.437(1), Pd-C1L, 2.026(6), PdBr 2.4484(7), C7-N1 1.12(1); P1-Pd-P2 92.38(5), P2-Pd-C1L 175.6(2), P1-Pd-C1L 91.0(2), P1-Pd-Br 173.51(4), P2-Pd-Br 90.90(4), C1L-$\mathrm{Pd}-\mathrm{Br} 85.4(2)$.
$\mathrm{Pd}-\mathrm{P} 2$ bond, trans to the aryl, at 2.437(1) $\AA$ represents the longest $\mathrm{Pd}-\mathrm{P}$ bond ever reported. $\S$ For comparison, in the Binap $\beta$-pinene allyl complex $\left[\operatorname{Pd}\left(\eta^{3}-\mathrm{C}_{10} \mathrm{H}_{15}\right)\{(R)-(+)\right.$-Binap $\left.\}\right]$ $\left[\mathrm{CF}_{3} \mathrm{SO}_{3}\right], 3$, the $\mathrm{Pd}-\mathrm{P}$ distances are 2.312(3) $\AA$ and 2.347(5)


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$\AA .{ }^{15}$ The observed $\mathrm{Pd}-\mathrm{P} 2$ bond in 1, 2.3501(9) $\AA$, is much shorter, but lies towards the upper end of the literature range (Pd-P separations of the order of $c a .2 .20-2.36 \AA$ are common ${ }^{16-19}$ ). In both $\mathbf{1}$ and 2 the $p$-CN group appears to be a normal triple bond (see captions).

Despite the superficial similarity of the ligands, i.e., biarylbased triaryl phosphine types, the structures for $\mathbf{1}$ and $\mathbf{2}$ are very different. Whereas the extreme lengthening of the $\mathrm{Pd}-\mathrm{P} 2$ bond in the Binap complex 2 arises due to good donor properties of the aryl group, the MeO-Biphep analog 1 avoids this electronic strain by strongly deviating from square planar geometry and thus weakening the $\mathrm{Pd}-\mathrm{C} 1 \mathrm{~L}$ overlap. Given the rather novel structural results for these $p-\mathrm{NCC}_{6} \mathrm{H}_{4}$ aryl compounds, it is not surprising that we cannot readily isolate analogous complexes with aryl ligands which are even more electron donating, e.g. $p-\mathrm{MeC}_{6} \mathrm{H}_{4}$ and $p-\mathrm{MeOC}_{6} \mathrm{H}_{4}$.

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## Notes and references

$\dagger$ The complexes were prepared as described in reference 11.
$\ddagger$ Crystal data for compound 1: $\mathrm{C}_{45} \mathrm{H}_{36} \mathrm{BrNO}_{2} \mathrm{P}_{2} \mathrm{Pd}, \quad M=871.00$, orthorhombic, space group $P 2_{1} 2_{1} 2_{1}$ (no. 18), $a=11.4485(2), b=$ 17.3604(1), $c=19.0940(3) \AA, U=3794.94(9) \AA^{3}, Z=4, \mu=16.66 \mathrm{~cm}^{-1}$, $T=200 \mathrm{~K}, R 1=0.0267$ (for 5794 unique reflections having $I>2 \sigma(I)$ ), 0.0354 (for all 6492 independent reflections). Crystal data for compound 2. $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ : $\mathrm{C}_{56} \mathrm{H}_{46} \mathrm{BrCl}_{2} \mathrm{NP}_{2} \mathrm{Pd}, M=1052.16$, monoclinic, space group $C 2$ (no. 5), $a=29.0943(4), \quad b=11.8671(2), c=17.2953(2) \AA$, $\beta=116.790(3)^{\circ}, U=5330.51(13) \AA^{3}, Z=4, \mu=12.94 \mathrm{~cm}^{-1}, T=293 \mathrm{~K}$, $R 1=0.0419$ (for 5550 unique reflections with $I>2 \sigma(I)$ ), 0.0494 (for all 7049 independent reflections). CCDC reference number 186/1478.
$\S$ Note added at proof: a Pd-P bond length of ca. $2.5 \AA$ (trans to $\mathrm{SiCl}_{3}$ ) has been observed. ${ }^{21}$

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